BLITZ: Nuclear Form I-L

Name	Period
This is a Take Home Exam. You may use your notes but you may NOT use help from human be	eings.
EXPLAIN IN COMPLETE SENTENCES AND GIVE EXAMPLES:	
You MUST <u>HAND WRITE</u> THIS EXAM!! NO TYPED PAPERS WILL BE ACCEPTED!	
1. Define alpha, beta, and gamma rays, and using a diagram tell how they were discovered.	
2. Using a diagram, explain how the charge-mass ratio of an electron is determined.	
3. Explain nuclear fission and fusion and give an example of each.	
4. Using a diagram, explain Isotopes and how they were discovered.	
5. Using a diagram, describe how ions and the proton were found.	
6. Discuss the meaning of half-life, and give an example.	
7. Using a diagram, tell the function of the five parts of a nuclear reactor.	
8. Using a diagram, tell how X-Rays are produced and two things they show us about atomic stru-	ucture.
9. Explain critical mass and the chain reaction.	
10. List five properties of radioactivity.	
*** COPY THESE EQUATIONS AND COMPLETE THEM ON YOUR PAPER:	
11. $_{99}\text{Es}^{254}$ + $_{2}\text{He}^{4}$ > ? + $_{0}n^{1}$	
12. ${}_{6}C^{12}$ + ?> ${}_{102}No^{254}$ + 2 ${}_{0}n^{1}$	
13. ${}_{94}Pu^{239} + {}_{0}n^1 - ?$	
14. $_{1}H^{2} + _{1}H^{3}{2}He^{4} + ?$	
15. $_{2}\text{He}^{4} + _{13}\text{Al}^{27}> _{14}\text{Si}^{30} + ?$	
16. ${}_{1}H^{2} + {}_{6}C^{12}> {}_{7}N^{13} + ?$	
17. ? + $_0n^1$ > $_{94}Pu^{241}$	
18. ${}_{5}B^{11} + {}_{94}Pu^{251}> ? + 3 {}_{0}n^{1}$	
19. $_{92}U^{238}$ + ?> $_{92}U^{239}$	
20. $_{3}Li^{6} + _{0}n^{1} - ? + _{1}H^{3}$	
When finished please STAPLE this evan onto your papers and turn in on due date	

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