Nuke Exam 3/20/05 2:08 PM

BLITZ: Nuclear Form M-R

Name	Period	

This is a Take Home Exam. You may use your notes but you may NOT use help from human beings.

EXPLAIN IN COMPLETE SENTENCES AND GIVE EXAMPLES:

You MUST HAND WRITE THIS EXAM!! NO TYPED PAPERS WILL BE ACCEPTED!

- 1. Explain critical mass and how to make an A-Bomb.
- 2. Using a diagram, explain how Ions and the Proton were discovered.
- 3. Discuss the meaning of half-life, and give an example.
- 4. Using a diagram, explain how Isotopes were discovered.
- 5. Give five properties of radioactivity.
- 6. Using a diagram, explain how the charge to mass ratio of an electron is determined.
- 7. Describe the diffusion method of separating isotopes.
- 8. Explain nuclear fission and fusion and give an example of each.
- 9. Using a diagram, tell the function of the five parts of a nuclear reactor.
- 10. Define alpha, beta, and gamma rays and, using a diagram, tell how they were discovered.

*** COPY THESE EQUATIONS AND COMPLETE THEM ON YOUR PAPER:

11.
$$? + {}_{0}n^{1} - --> {}_{94}Pu^{241}$$

12.
$$_{5}B^{11} + _{04}Pu^{251} ---> ? + 3_{0}n^{1}$$

13.
$$_{3}Li^{6} + _{0}n^{1} - > ? + _{1}H^{3}$$

$$14._{04}Pu^{239} + _{0}n^{1} - > ?$$

15.
$$_{99}Es^{254} + _{2}He^{4} ---> ? + 2_{0}n^{1}$$

16.
$$_{6}C^{12} + ? \longrightarrow _{102}No^{254} + 2_{0}n^{1}$$

17.
$$_{1}H^{2} + _{1}H^{3} ---> _{2}He^{4} + ?$$

18.
$$_{92}U^{238} + ? ---> _{92}U^{239}$$

19.
$$_{2}\text{He}^{4} + _{13}\text{Al}^{27} ---> _{14}\text{Si}^{30} + ?$$

20.
$$_{1}H^{2} + _{6}C^{12} - > _{7}N^{13} + ?$$

When finished, please STAPLE this exam onto your papers and turn in on due date.